

the  $g$  tensor components reported here and those of the pure [CuMg] complex<sup>8</sup> is found to be good, particularly in view of the uncertainty of the measurement of  $g_{\parallel}$  in the powder spectrum of the pure material.

[CuNi]. The [CuNi] spectra (Figure 3, full lines) are much more difficult to analyze than those of [CuMg], since the resonant fields of the three magnetically inequivalent binuclear sites are distinct. Again, hyperfine structure is present, splitting each resonance of the  $k_3k_1$  and  $k_1k_2$  planes into four equidistant lines. Because of the very low abundance (1.19%) of the <sup>61</sup>Ni isotope with  $I = 3/2$ , this hyperfine structure arises exclusively from the coupling of the  $S = 1/2$  electronic ground state with the  $I = 3/2$  nuclear spin of copper. The contributions of the three magnetically inequivalent sites are clearly displayed in the  $k_2k_3$  plane, whose normal was found to be only  $17^\circ$  from the trigonal axis. A total of twelve lines were expected in this plane, but only three broad envelopes are resolved, in agreement with the very weak value of  $A_{\perp}$  (see below). The three envelopes are out of phase by  $60^\circ$  and two of them intersect every  $30^\circ$ . When  $H$  is parallel to  $k_3$ , the intensity of the high-field line is about twice that of the low-field line; two of the three magnetic sites, with their molecular  $x$  axes  $30^\circ$  from  $H$ , contribute to the high-field line, while only one magnetic site, with its molecular  $y$  axis parallel to  $H$ , corresponds to the low-field line. Inversely, when  $H$  is parallel to  $k_2$ , the low-field group of lines, which now represents the contributions of the two magnetic sites with their molecular  $y$  axes approximately  $30^\circ$  from  $H$ , becomes more intense than the high-field group. In the  $k_3k_1$  and  $k_2k_1$  planes, where  $H$  is rotated from the  $k_3$  or the  $k_2$  axis toward the  $c$  axis, eight hyperfine lines collapse into four equally intense lines, as expected.

The  $g$  and  $A$  parameters deduced from the analysis of the angular dependence of the  $g^2$  and  $g^2A^2$  tensors for each of the three magnetic sites are given in Table I. The minimum principal  $g$  value is found along the  $c$  axis; its value is 2.22, in excellent agreement with that determined from the study of the pure compound.<sup>8</sup> In contrast, whereas the  $g$  tensor of the pure compound appears to be axial, in the diluted compound it exhibits a rhombic symmetry with  $g_x = 2.25$  and  $g_y = 2.36$ , where the principal magnetic axes coincide with the molecular axes  $x$  and  $y$ . The maximum principal  $A$  value, equal to  $69 \times 10^{-4} \text{ cm}^{-1}$  is found along the  $c$  axis. In the  $ab$  plane, the hyperfine parameter is found smaller than  $10 \times 10^{-4} \text{ cm}^{-1}$  and cannot be determined with accuracy.

The appropriate spin Hamiltonian to describe the low-lying states in [CuNi] is

$$\mathcal{H} = \beta H \cdot (g_{\text{Cu}} \hat{S}_{\text{Cu}} + g_{\text{Ni}} \hat{S}_{\text{Ni}}) + \hat{S}_{\text{Ni}} \cdot \mathbf{D} \cdot \hat{S}_{\text{Ni}} - J \hat{S}_{\text{Cu}} \cdot \hat{S}_{\text{Ni}} + \hat{S}_{\text{Cu}} \cdot \mathbf{D}_{\text{CuNi}} \cdot \hat{S}_{\text{Ni}} + \hat{I}_{\text{Cu}} \cdot \mathbf{A}_{\text{Cu}} \cdot \hat{S}_{\text{Cu}} \quad (1)$$

where the meaning of the symbols, but the last one, has already been specified.<sup>8</sup> The last term is the copper(II) single-ion hyperfine structure term. The spin Hamiltonian associated with the  $S = 1/2$  ground pair state is

$$\mathcal{H}_{1/2} = \beta H \cdot g_{1/2} \hat{S} + \hat{I}_{\text{Cu}} \cdot (\mathbf{A}_{\text{Cu}})_{1/2} \cdot \hat{S} \quad (2)$$

where the meaning of the symbols is obvious. The relations between  $g_{1/2}$ ,  $g_{\text{Cu}}$ , and  $g_{\text{Ni}}$ , on the one hand,  $(\mathbf{A}_{\text{Cu}})_{1/2}$  and  $\mathbf{A}_{\text{Cu}}$ , on the other hand, are

$$g_{1/2} = \frac{4g_{\text{Ni}} - g_{\text{Cu}}}{3} \quad (3)$$

$$(\mathbf{A}_{\text{Cu}})_{1/2} = -\frac{1}{3} \mathbf{A}_{\text{Cu}} \quad (4)$$

We already discussed the validity of relation 3.<sup>8</sup> As far as the hyperfine interaction is concerned, the validity of relation 4 is nicely checked for the well-determined principal value of the  $A$  tensor, since  $(A_{\parallel\text{Cu}})_{1/2} = 69 \times 10^{-4} \text{ cm}^{-1}$  and the corresponding value in the same direction for [CuMg] is  $A_{\text{Cu}} = 222 \times 10^{-4} \text{ cm}^{-1}$ . We find the agreement to be quite good<sup>12</sup>

and definitely better than that found in two similar single-crystal studies.<sup>2,11</sup> The fact that the smallest value of the hyperfine coupling could not be established is also a check of relation 4 since the perpendicular value for the pair estimated from the corresponding one in [CuMg] should be less than  $10 \times 10^{-4} \text{ cm}^{-1}$ .

The agreement found between the observed hyperfine parameters in [CuNi] and [CuMg] and those deduced from the spin Hamiltonian (1) suggests that the approximations on which it is based are valid.<sup>8</sup> These approximations are as follows: (i) the wave functions of the real electrostatic Hamiltonian (including spin-orbit coupling and Zeeman terms) appropriate to describe the low-lying pair states are Heitler-London functions constructed from eigenfunctions of local Hamiltonians; (ii) the coupling between ground configuration ( $\text{Cu}^{\text{II}}\text{Ni}^{\text{II}}$ ) and metal-metal charge-transfer configuration ( $\text{Cu}^{\text{III}}\text{Ni}^{\text{I}}$  or  $\text{Cu}^{\text{I}}\text{Ni}^{\text{III}}$ ) is not taken into account. Furthermore, we can specify the range of validity of the relations (3) and (4). In [CuNi], the isotropic exchange is found to be of medium magnitude. But if  $|J|$  were very small, the total spin  $S$  would not be a good quantum number any longer and the meaning of the Hamiltonian (2) would vanish. At the other extremity, if  $J$  were very negative, the Heitler-London description of the low-lying states, which is implicitly required by the spin Hamiltonian (1), would not be appropriate any longer.

Since we did not observe any EPR transition within the upper quartet state of [CuNi], we were not able to check relations relative to zero-field splitting effects, similar to (3) and (4), as has been done with other heterobinuclear systems.<sup>4,5,7</sup> In order to specify the range of validity of the approximations required by the spin-Hamiltonian formalisms (1) and (2), we shall study other Cu(II)-Ni(II) pairs with smaller doublet-quartet energy gaps. To conclude, we want to emphasize that the [CuNi] complex considered in this note is apparently the first heteropair for which we have both the energy gap between low-lying spin states owing to accurate magnetic measurements on the pure material and the details of the electronic structure of the ground state owing to this EPR study on a doped diamagnetic host lattice. We also wish to point out that [CuCo] could be used as EPR-silent host lattice for investigating the properties of low-lying states in other pairs.<sup>13</sup>

**Acknowledgment.** S.R.D. thanks the CNRS for a post-doctoral appointment in 1981-1982.

**Registry No.** CuCo(fsa)<sub>2</sub>en(H<sub>2</sub>O)<sub>2</sub>, 62109-46-0; Mg, 7439-95-4; Ni, 7440-02-0.

(12) One of the reviewers suggested that an additional term in the Hamiltonian (1) of the form  $\hat{I}_{\text{Cu}} \cdot \mathbf{A}'_{\text{Cu}} \cdot \hat{S}_{\text{Ni}}$  could account for the weak deviation with regard to relation 4.

(13) In fact, making use of this conclusion, we also prepared [CuCo] doped with Mn with an initial Mn:Co ratio of 1:20. Unfortunately, the EPR spectrum of the resulting well-shaped crystals proved to be of prohibiting complexity.

Contribution from the Department of Chemistry, College of Arts and Sciences, The University of Tokyo, Komaba, Meguro-ku, Tokyo 153, Japan, and ASAI Germanium Research Institute, Izumihoncho, Komae-shi, Tokyo 201, Japan

#### Germanium-73 Chemical Shifts and Spin-Lattice Relaxation Times of Some Tetrasubstituted Germanes

Yoshito Takeuchi,<sup>\*,†</sup> Toshie Harazono,<sup>†</sup> and Norihiro Kakimoto<sup>†</sup>

Received January 9, 1984

As compared with NMR spectroscopy of other group 4B elements, that of <sup>73</sup>Ge has attracted less attention chiefly

Table I.  $^{73}\text{Ge}$  NMR Parameters for Tetrasubstituted Germanes of the Type  $\text{RX}_4$  or  $\text{RX}_3\text{X}'$ 

compd	X	$\delta(\text{Ge})^a$	$\nu_{1/2}/\text{Hz}$	$T_1^b/\text{ms}$ (temp/K)
1	Me <sup>c</sup>	0.0	1.3	378 (343)
				313 (315)
				295
				270 (277)
				223 (255)
2	Et <sup>c</sup>	17.3	1.7	208 (240)
				274 (323)
				250
				175 (279)
				135 (255)
3	Ph <sup>d,e</sup>	-31.6	6	287
				67 (214)
4	2-furyl <sup>d</sup>	-112.8	11	20
5	2-thienyl <sup>d,e</sup>	-95.5	8	44
6	Cl <sup>b</sup>	30.9	1.2	287
7	Me <sup>f</sup>	6.3	54	

<sup>a</sup> Relative to external tetramethylgermane. <sup>b</sup> At 303 K unless specified. <sup>c</sup> In  $\text{CDCl}_3$ . <sup>d</sup> In  $\text{Me}_2\text{SO}-d_6$  (saturated solution).

<sup>e</sup> At 329 K. <sup>f</sup>  $\text{X}' = \text{CH}_2\text{CH}_2\text{COOH}$ .

because of its electric quadrupole moment that tends to cause an excessive broadening of  $^{73}\text{Ge}$  resonances,<sup>1</sup> and so far, only a limited number of  $^{73}\text{Ge}$  resonances have been recorded for compounds of the type  $\text{GeR}_4$  or  $\text{GeR}_3\text{R}'$ , where R or R' is an alkyl, a substituted alkyl, or a halogen.<sup>2-4</sup> Here, we will describe chemical shifts and spin-lattice relaxation times ( $T_1$ ) of some tetrasubstituted germanes.

In Table I,  $^{73}\text{Ge}$  NMR parameters of a variety of symmetrical molecules, tetramethyl- (1), tetraethyl- (2), tetraphenyl- (3), tetrakis(2-furyl)- (4), tetrakis(2-thienyl)- (5), and tetrachlorogermanes (6) are listed. As an example of an unsymmetrical germane,  $\beta$ -(trimethylgermyl)propionic acid (7) was included. The symmetrical species 1-6 gave reasonably sharp signals ( $\nu_{1/2} = 1.2-11$  Hz) while 7 has  $\nu_{1/2}$  of 54 Hz. It was reported that for unsymmetrical species of the type  $\text{GeR}_3\text{R}'$ ,  $\nu_{1/2}$  ranges from 20 to 300 Hz.<sup>4</sup> This broadening is expected to reflect, as is the case with  $^{14}\text{N}$  resonances, the electric field gradient.

For a variety of compounds, Zinmane et al.<sup>4</sup> reported a correlation given below between the  $^{73}\text{Ge}$  chemical shifts,  $\delta(\text{Ge})$ , and  $^{29}\text{Si}$  chemical shifts,  $\delta(\text{Si})$ , of compounds of the type  $\text{Me}_3\text{MR}'$  (M = Si, Ge), where R' is either substituted alkyl or alkenyl residues.

$$\delta(\text{Ge}) = 1.85 \delta(\text{Si}) + 1.52 \quad \text{rms} = 0.970$$

We attempted a similar correlation between  $\text{R}_4\text{Ge}$  and  $\text{R}_4\text{Si}$ , where R = Me, Et, Ph, and 2-furyl to obtain

$$\delta(\text{Ge}) = 2.01 \delta(\text{Si}) + 1.11 \quad \text{rms} = 1.00$$

It is interesting to notice that the slope is not very far from that reported by Zinmane et al.,<sup>4</sup> although in the latter correlation Ge nuclei bonded to  $\text{sp}^3$ - rather than  $\text{sp}^2$ -hybridized carbons are involved. In view of much abundant shift data for  $^{29}\text{Si}$ , this correlation will be of help in predicting  $^{73}\text{Ge}$  chemical shifts.

To the best of our knowledge there seems no systematic report on  $^{73}\text{Ge}$  spin-lattice relaxation times ( $T_1$ ) of organogermanium compounds although those of  $^{29}\text{Si}$ ,<sup>5</sup>  $^{117}\text{Sn}$ ,<sup>6</sup> and  $^{207}\text{Pb}$ <sup>7</sup> have been investigated to some extent. In Table I, the  $^{73}\text{Ge}$  and  $T_1$  values of 1, 2, and 4-6 determined by the inversion-recovery method are recorded. For 1 and 2,  $T_1$  values determined at various temperatures were also listed. A plot of  $\ln T_1$  as a function of  $1/K$  gave a straight line (rms = 0.989 for 1, 0.996 for 2) with a negative slope. This kind of temperature dependency of  $T_1$  values is often observed for cases where either the dipolar or quadrupolar relaxation is predominant. The former mechanism is not important for  $^{73}\text{Ge}$ , a quadrupolar nucleus. Indeed, we found that the NOE enhancement factor for 1 and 2 is null within the range of experimental error. Thus, the main pathway of relaxation is quadrupolar. If the spin-rotation relaxation is operative at any significant extent, the plot of  $\ln T_1$  vs.  $1/K$  cannot be a straight line for a wide range of temperature since this relaxation shows a reverse temperature dependency to that of dipolar or quadrupolar relaxations. All of our present observations indicate that  $^{73}\text{Ge}$  relaxes exclusively via quadrupole mechanism<sup>8</sup> within the range of experimental error.

### Experimental Section

$^{73}\text{Ge}$  NMR spectra were recorded on JEOL FX-90Q spectrometer equipped with a low-frequency insert operating at 3.10 MHz in a 10-mm tube at 30 °C. The typical conditions of measurement are as follows: pulse width, 150  $\mu\text{s}$  (90°); spectral width, 100-1000 Hz; number of scans, 100-1000; pulse delay, 50 ms; data points, 2048-8192.  $T_1$  values were determined by the inversion-recovery method with a minimum delay of  $10T_1$ : all the samples in  $\text{CDCl}_3$  were degassed by the freeze-thaw method.

The compounds used in this investigation are all known compounds except for 7<sup>9</sup> and were prepared according to the standard method.

**Note Added in Proof.** After the submission of our paper, we found a paper treating  $^{73}\text{Ge}$  relaxation in some alkyl- and alkoxygermanes: Sekatsis, I. P.; Liepins, E.; Zicmane, I. A.; Lukevics, E. *Zh. Obshch. Khim.* 1983, 53, 2064.

**Registry No.** 1, 865-52-1; 2, 597-63-7; 3, 1048-05-1; 4, 55811-78-4; 5, 62473-54-5; 6, 10038-98-9; 7, 33076-65-2;  $^{73}\text{Ge}$ , 15034-58-9.

- (5) Levy, G. C.; Cargioli, J. D.; Juliano, P. C.; Mitchell, T. D. *J. Am. Chem. Soc.* 1973, 95, 3445.
- (6) Bluden, S. J.; Frangou, A.; Gillies, D. G. *Org. Magn. Reson.* 1982, 20, 170.
- (7) Maciel, G. E.; Dallas, J. L. *J. Am. Chem. Soc.* 1973, 95, 3039.
- (8) Lehn, J. M.; Kinzinger, J. P. "Nitrogen NMR"; Witanowski, M., Webb, G. A., Eds.; Plenum Press: New York, 1973.
- (9) Prepared from  $\beta$ -(trichlorogermyl)propionic and methylmagnesium bromide.

Contribution from the Department of Chemistry,  
Faculty of Science, Ehime University,  
Matsuyama, Ehime 790, Japan,  
and Department of Applied Molecular Science,  
Institute for Molecular Science,  
Okazaki National Research Institutes, Okazaki 444, Japan

### Piezochromism of Nickel(II) Complexes with Tetraaza Macrocylic Ligands in Water

Yoichi Kitamura,\*<sup>1a</sup> Tasuku Ito,<sup>1b</sup> and Masako Kato<sup>1b</sup>

Received February 9, 1984

It is well-known that Ni(II) complexes with macrocyclic ligands (L) exist in water in an equilibrium between yellow,

(1) (a) Ehime University. (b) Okazaki National Research Institutes.

<sup>1</sup> The University of Tokyo.

<sup>1a</sup> ASAI Germanium Research Institute.

- (1) Harris, R. K.; Kennedy, J. D.; McFarlane, W. In "NMR and the Periodic Table"; Harris, R. K., Mann, B. E., Eds.; Academic Press: New York, 1978; p 309.
- (2) Kaufmann, J.; Sahn, W.; Schwenk, A. *Z. Naturforsch.*, A 1971, 26A, 1384.
- (3) Kidd, R. G.; Spinney, H. G. *J. Am. Chem. Soc.* 1975, 95, 88.
- (4) Zicmane, I.; Liepins, E.; Lukevics, E.; Gar, T. K. *Zh. Obshch. Khim.* 1982, 52, 896.